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X-ray Attenuation Coefficients From 10 kev to 100 Mev









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X-ray Attenuation Coefficients From 10 key to 100 Mey

Rosemary T. McGinnies



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National Bureau of Standards Supplement to Circular 583

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X-ray Attenuation Coefficients from 10 key to 100 Mey

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A revision is given of the X-ray attenuation coefficients presented in National Bureau of Standards Circular 583. Table 4 of that publication is eliminated, and a new table is given for each material for photon energies less than 100 key. The uncertainties in the estimates of attenuation coefficients at low energies are from 3 to 5 percent, which is the same as was previously given at higher energies. The cross sections for scattering are unchanged. Two values are listed for the photoelectric cross section, one calculated from the Sauter-Stobbe formulas and the other derived from new experimental evidence. The procedures for smoothing experimental data are described and are generally the same as were used in Circular 583. In addition to the systematic coverage of the region from 10 key to 100 Mey, some data are included for a number of elements based on experimental measurements below 10 key and above 100 Mey. A comparison is made between calculated and experimental total attenuation coefficients at energies above 10 Mey.

1. Introduction

The estimates of X-ray attenuation coefficients presented in National Bureau of Standards Circular 583 [1] were based on a combination of theoretical calculations and available experimental evidence, with judicious interpolations between them. Inaccuracies in the values below 50 kev were thought to approach 10 percent, especially for light elements, but probably not to exceed 3 to 5 percent above 100 key. Experimental data obtained in the past few years enable us to revise the estimates at low energies so that they are of the same accuracy as the others. The main tables in this revision are intended to replace the low-energy portions of the tables in Circular 583. The footnotes to these tables are omitted even though they still apply. columns giving the cross sections for coherent and incoherent scattering are unchanged. The same general procedures are used here as in the original publication for the energy region from 10 key to 100 Mey. In addition, we include some data for a number of elements based on experimental measurements below 10 key and some information up to 1 Bev. No attempt is made to break down these total attenuation coefficients into contributions from individual absorption and scattering processes.

No revision is required at this time for the intermediate energy region from 100 kev to 20 Mev. There are no new measurements of total attenuation coefficients at these energies. The estimates just below 1 Mev have been substantiated to within an experimental error of 2 percent in recent direct measurements of the photoelectric cross sections of Cu, Mo, Ag. Ta, and Au for Cs¹³⁷ radiation [2]. A direct measurement of the photoelectric cross section for Pb at 0.511 Mev is also in reasonable agreement with the estimates already given [3].

2. Low Energies ($h\nu < 40$ kev)

Table 1 shows the range of recent good measurements of X-ray attenuation coefficients at energies below 40 kev [4, 5, 6, 7, 8]. The accuracy claimed by each experimenter is listed in table 2. Most of the data are due to Alan J. Bearden and R. D.

The empirical corrections to calculated photoelectric cross sections at energies below 100 key for low-Z materials given in table 4 of Circular 583 were based on a small number of experimental data, all of which were obtained by obsolete techniques. This low-energy region has now been studied quite extensively below 30 or 40 key by modern experimental methods [4, 5, 6, 7, 8]. It appears that calculations based on the Sauter-Stobbe formula are in somewhat better accord with experiment for low-Z materials and for 10 to 40 kev photons than was previously indicated. On the other hand, data at low energies for the high-Z materials require considerable revision of attenuation coefficients below the K absorption edges. The reason for this is that a hydrogen-like approximation, such as is made in the Stobbe formulas, is not nearly so good for L and M shells as for the K shell. Increases ranging from 17 to 38 percent are indicated for Mo, Sn, and I and from 40 to over 100 percent for the very-high-Z elements ($Z \ge 74$). The estimates for elements Mo through U below the K edges are the only ones for which the accuracy of the estimates given in Circular 583 falls below the stated amount.

Considerable new experimental data are also now available above 20 Mev [9, 10, 11, 12]. There is a general trend in the region below 100 Mev for the experimental values of the total attenuation coefficients in low-Z elements to be several percent higher than estimated in Circular 583. We give here a brief account of recent developments and indicate attempts that have been made to understand the measurements. A revision of the estimates of Circular 583 at higher energies will be given in a future publication when, it is hoped, the theoretical picture may be clarified.

Deslattes who claim estimated uncertainties of less than 1 percent [4, 5]. Each will give the details of his work in a forthcoming paper in the *Physical Review*; Bearden is currently extending his measurements below 10 key to other elements

and somewhat lower energies. French measured mass-attenuation coefficients for Cu and Al down to 13.2 kev; Hopkins extended these data to 7.5 kev and to the other elements listed in the table [6]. The data of Roof for Th, U, and Pu are consistent with those available for W, Pt, and Au; however most of his measurements for the light elements at 17.48 kev (Mo $K_{\alpha 1}$ radiation) depart substantially from those of other experiments [7].

The experimantal coverage is far from complete so that extensive interpolation with respect to energy and atomic number is required for a systematic tabulation of attenuation coefficients. For each element listed in table 1, all the experimental data were plotted on a large graph as functions of energy. The critical absorption edges were located at the energies listed in table 1. A smooth curve was drawn through the points between edges. Values read off this curve at 3, 4, 5, 6, 8, 10, 15, 20, and 30 kev were converted to cross sections per atom by multiplying by A/N_A , where N_A is Avogadro's number and A is the atomic weight. At each energy the cross sections per atom should be smooth functions of Z so that one can interpolate for missing elements. The two-way smoothing versus energy and Z was done by successive approximations for different groups of data. At a particular energy, starting with the lightest elements, those whose K edges lie at lower energies form one group, those whose K edges lie at higher energies and L edges at lower energies form another, and so on. For example, at 3 key elements below Cl at Z=17 form one group and at 10 key elements below Z=30 (Zn) form another.

There have been no calculations of the effect of binding on scattering cross sections below 10 key. Therefore, in this region the total cross sections per atom were smoothed graphically as functions of Z. The fitting was aided by log-log plots of the attenuation coefficients versus energy, which form straight lines at 10 key and below. The results below 10 key given in table 3 fit smoothly with the higher energy values in the other tables.

More precise smoothing of the data was possible at 10, 15, 20, and 30 key. The three groups of interest which the elements form for these energies are those with $Z \leq 30$, $40 \leq Z \leq 53$, and $Z \geq 74$. (Elements with Z between 53 and 74 are omitted.) The total scattering (coherent plus incoherent) corrected for binding effects, which is given both here and in Circular 583 in the second column of the table for each element, was subtracted from the experimental cross sec-

tion per atom to give an experimental photoelectric cross section. For $Z \leq 30$, the relative deviation of this quantity from the photoelectric cross section calculated by the Sauter-Stobbe formulas was smoothed as a function of Z by a least-squares fit for a straight line. The smoothed experimental photoelectric cross sections are given in column 5 of tables 4 through 29 and are used to calculate the totals in columns 6 and 7; for comparison. the calculated photoelectric cross sections are listed in column 4. The new experimental data for Zr (40) through Sn (50) were used to revise the tables for Mo, Sn, and I. At each energy the logarithm of the experimental photoelectric cross section is a good linear function of Z in this region so that a straight-line fit by the method of least squares was determined. These lines were extrapolated to Z=53 to obtain new values for I. Due account was taken of the absorption edges and log-log plots of the adjusted photoeffect cross sections as functions of energy aided in completing the tables. There is less than 2-percent difference between any experimental point and the corresponding point obtained by the least-squares fit. Below the K edges the new values for the mass-attenuation coefficients for these elements are 20 to 40 percent higher than those calculated from the Stobbe formulas. Perhaps this straightline extrapolation is not good. Incidentally, the new values do agree well with those compiled by Allen for Compton and Allison's book [14].

The new estimates of the mass-attenuation coefficients for the heaviest elements $(Z \ge 74)$ depend upon the experimental data for W, Pt, Au, Th. U, and Pu [4, 5, 7], which are not very complete. The L-absorption edges break up the region so that smoothing versus Z is possible only at 10, 15, and 30 key. This was done graphically by means of a semilog plot of the experimental photoelectric cross section (in barns per atom) as a function of Z. Values for Tl and Pb interpolated from these curves were added to the total scattering cross sections, and the totals were plotted on log-log paper as functions of energy for each element. These form straight lines between edges. The results of this two-way smoothing and interpolating are given in tables 22 through 26. The agreement with Allen's values is very good for Pb, good for W and Pt away from the edges, fair for Pt, and poor for Th. Although the new estimates may be too high, they are based on the only experimental values that are available. The experimental procedure of reference [7] will be discussed in reference [4].

3. Intermediate Energies (30 < hv < 100 kev)

The only new experimental data at energies above 30 kev are for Al. Cu, Sn, and Au at 40 Kev. These data together with the new values at 30 kev for all Z must be made consistent with the older data of high accuracy mentioned in Circular 583. Over 20 years ago, Cuykendall [15] measured low-Z materials at energies between 60 and 250

kev with probable errors no greater than 2 percent, and Jones [16] covered the high-Z range at energies from 67 to 447 kev with an average probable error of 0.8 percent. There is nothing of comparable accuracy between 40 and 60 kev. A comparison between Cuykendall's low-Z data and predictions based on the Sauter-Stobbe formulas

shows that the experimental values are from 1 to 10 percent higher than the calculated ones, that this difference goes to zero at about 100 key, and that for Cu the difference is zero at all these energies. Cuykendall's data from 60 to 90 key are larger relative to the calculated values than the newer data are at about 35 key. However, the uncertainty in the calculations is about the same from about 35 to 100 key, provided the entire region is far above the K-absorption edges. Therefore, the current estimates for the attenuation

coefficients for elements with Z<30 are obtained by smoothing between the predictions of the Sauter-Stobbe formulas above 40 keV and the recent experimental values below that energy. Jones' data for Mo and Sn above about 70 keV agree with the estimates given in Circular 583. There are no new experimental data for these materials above 30 keV so that calculated values are assumed above the K edges. Jones' data for Pb in the vicinity of the K edge agree with the revised values given in table 25.

4. High Energies (hv > 10 Mev)

Attenuation coefficients for a range of elements at energies above 10 Mey are given in figure 1. The solid curves are the totals estimated in Circular 583, and the points are the experimental data now available. The latter are quite consistent and indicate clearly that for low-Z materials the values based on the theory assumed in Circular 583 are too low by several percent. This discrepancy was indicated in Circular 583 but not considered to be serious because of the limited experimental information available at that time and the general difficulty of assigning a definite cross section to each of the various processes that contributes to the total attenuation coefficient.

Present evidence confirms the earlier trend and points to a substantial problem above 10 Mev. Wolff measured the total attenuation coefficients of C and H₂O at 20.3 and 20.8 Mev, which is in the region of strong photonuclear absorption [12]. Wyckoff and Koch made very extensive and accurate measurements for C, Al, H₂O, and H from 13 to 82 Mev [12]. The other new data below 100 Mev were obtained by Moffatt, et al., at 94 Mev for a range of elements from H to U [10]. Information about high-Z elements is still not adequate. The measurements above the photonuclear region (35 Mev) for C, Al, and H₂O are especially accurate with maximum uncertainties of about 0.9 percent [10, 11].

It is unclear at this time what the best estimates are for calculated values of the cross sections for individual processes at high energies, particularly for electron pair production in the field of the atomic electrons. In the first report of this series [28], calculations from the Borsellino results for free electrons were used up to 50 Mev [29], with an extrapolation to higher energies aided by the results of Wheeler and Lamb [30].

In Circular 583, calculations of the pair cross section in the field of electrons were made by using the formulas of Votruba for the limiting cases of photon energies near the threshold value and large compared to mc² [31]. A graphical interpolation was made for the intermediate energy region. This was accomplished by extrapolating the curves for the limiting cases and using the calculations of Borsellino [29] as a guide to the shape of the curve in the intermediate region. This general

procedure was recommended by Rohrlich. More recently he and Joseph re-evaluated the exchange correction in the high energy limit showing that the constant in eq 13, Circular 583, is exactly 100/9 instead of 11.3 ± 0.5 [33]. In the case of hydrogen, this correction raises the cross section for pair production in the field of the electrons at 30, 40, 60, 80, and 100 Mey by 0.1 mb and at 50 Mev by 0.2 mb. Neglecting the effect of binding on the atomic electrons, revised values of this cross section for other elements are obtained by multiplying that for hydrogen by Z. The same result is obtained above 20 Mev by numerical integration of the formulas of Wheeler and Lamb [30] for hydrogen over the energy of the positron and subtraction of the exchange correction of Joseph and Rohrlich [33].

Recently Suh and Bethe, following the work of Borsellino, have studied the theory of electron pair production in the field of a particle of arbitrary mass [29, 35]. The recoil-momentum-distribution function for high-incident photon energies which they obtained for the case of a recoil electron originally bound in an atom agrees with that of Wheeler and Lamb [30]. Measurements of electron pair production in the nuclear field and in the electron field in a hydrogen-filled cloud chamber by Hart, Cocconi, et al., between 10 and 100 Mey are also in agreement with the Wheeler-Lamb calculations [34]. Malamud compares the results of accurate experiments above 40 Mev with calculations of total attenuation coefficients at these energies [9]. He finds that measurements in Be and C clearly agree with the Wheeler-Lamb prediction and disagree with the Wheeler-Lamb result reduced by the exchange correction of Joseph and Rohrlich. Measurements in H and Li are inconclusive.

An experiment was designed in this laboratory to decide whether the discrepancy between experiment and current estimates should be assigned to nuclear or electronic effects [11]. The result of this was that even though an upper limit was taken for pair production in the electron field, the nuclear pair cross section was still too low by 2.25 percent (for C and Al). An exploration was made to see what the effect would be of replacing Thomas-Fermi form factors by Hartree form factors in the screening calculation for nuclear pair

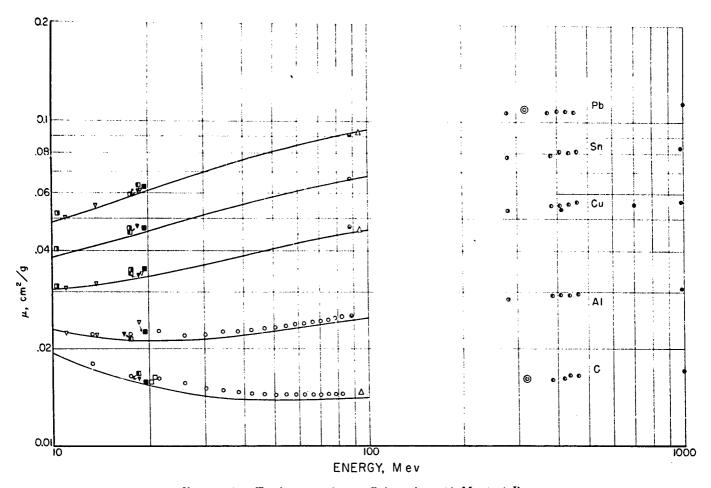


Figure 1. Total attenuation coefficients from 10 Mev to 1 Bev.

O Anderson, et al. [19, 20] Cooper [12]
Rosenbium [22] Walker [23] Colgate [24]

The solid curves are the totals given in NBS Circular 583. The points are experimental data from the following sources:

 Malamud [9]
 △ Moffatt, Thresher, Weeks, Wilson [10]
 ○ Wyckoff and Koch [11]
 □ Wolff [12] Lawson [17]
Dewire, Ashkin, Beach [18]

Berman [25] Adams [26] the attenuation coefficients at high energies seems unwarranted at this time.

production and also replacing the high-energy approximation made in Bethe and Heitler's formula 114, reference [27], by the exact no screening expression, eq 110, reference [27]. A sample calculation for C and Al at 60 Mev shows that these two effects nearly cancel so that the net result is to decrease the pair cross sections given in Circular 583 by 0.7 percent.

In view of these considerations, a revision of

The author thanks the many scientists who have sent her reports on their research and preprints of their publications and the members of this laboratory who have helped her by contrib-

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Table 1. Sources of low-energy experimental data

Element	Z.	Experi-	Limits of en	ergy range	Number of measure-	•	Absorption edge	× of interest =	
		menters			ments	к	L_1	L_2	L_3
le Se Se	2 4 6	a a e a c a b	ker 3 3 8 8 3 7 7	ker 8 8 17 8 17 8	6 6 2 6 3 0	ken	ker	ker	ker
\1 \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	13	a b c e a	3 8 7.5 7	40 30 40 17 8	15 19 19 3 6	3. 203			
Fi Or Fe Co	22 24 26 27	ի « « ի «	8 8 6 8 6	30 - 8 14 30 14	23 1 9 20 10	4, 964 5, 988 7, 111 7, 700			
Ni Cu	28 29	b c e a b	8 6 8 8 8	30 14 8 40 30 40	22 9 1 9 21 24	8, 331 8, 980			
Zn Zr Mo Pd Ag	30 40 42 46 47	c b b b	6 11. 2 8. 0 8	14 30 30 30 30	10 13 20 19	9, 660 17, 998 20, 002 24, 347 25, 517			
Cd Sn	48 50	b a b c e	8 8 11 20 8	30 40 30, 5 35 11	20 9 14 9 2	26, 712 20, 190			
W Pt Au Th	74 78 79 90	b a b d	8 8 8 5.4	30 30 40 30 25	18 19 9 19 12	69, 508 78, 379 80, 713	12, (84) 13, 873 14, 353 20, 460	11, 535 13, 268 13, 733 19, 688	10, 198 11, 559 11, 919 16, 296
U Pu	92 94	d d	5. 4 5. 4	25 25	12 12		21, 753 23, 097	20, 943 22, 262	17. 163 18. 066

[.] S. Fine and C. F. Hondee, Nucleonics 13, 39 (1955).

Table 2. Accuracy claimed by experimenters at low energies

a. Alan J. Bearden [4]
Probable error < 1.0 percent
b. Deslattes [5]
Standard deviation ≤ 1.0 percent
c. Horkins and French [6]
Probable error about 5 percent
d. Roof [7]
Probable error ≤ about 3 percent
c. Hubbell [8]
Probable error about 2 percent

Table 3. Total X-ray attenuation coefficients at photon energies below 10 kev (with coherent scattering) in cm2/g. (When two values are given for the same energy, they represent the maximum and minimum values at a critical absorption limit.)

Photon energy ker Element	3	3,20 *	4	5	6	8
Be	22, 5		9, 20	4, 65	2, 70	1, 10
C	89, 0		37, 5	18, 6	10, 0	4, 3
N	147, 4		63, 3	31, 7	17, 2	7, 40
O	221		100	50, 5	29, 0	11, 8
Ne	415		182	97, 0	50, 7	22, 3
Al	762	54	351	185	112	52, 3
A	200	1710	956	500	295	126

Photon energy kev	5,988 1	6,00	7.709 °	5,00	8,3314	8,96 r	9,00	9,660 f	10,00
Mg Ti Cr Fe Co	518	92	4-1 3tr2	40, 2 205 258 322 360	- 		28, 5 149 195 238		21. 1 115 146 180
NI Cu Zn		107 116		46 51, 5 61, 7	38. 5	37. 0 306	280 296	35. 3 265	210 225 240
Pd Ag Cd Sn W				204 217 230 258 177			149 160 170 188 128		112 123 120 147 93. 4
Pt Au Th U Pu		808 820 1080		201 208 400 407 510			147 155 300 306 370		115 122 219 244 278

Table 4. Beryllium

	Scatt	Scattering		electric	Total experiment		
Photon energy	With coherent	With- out coherent	Calculation K & L shells	Experi- ment	With coherent	With- out coherent	
Mer 0.01 .015 .02 .03 .04 .05 .06	Barns/ alom 3. 54 3. 01 2. 77 2. 53 2. 38 2. 29 2. 21 2. 10	Barns/ atom 2, 56 2, 52 2, 47 2, 39 2, 31 2, 24 2, 18 2, 07	Burns/ atom 4.75 1.25 0.48 .12 .05 .02 .01	Barns/ atom 5, 23 1, 34 0, 51 , 13 , 05 , 02 , 01	cm ² /y 0. 586 .201 .210 .178 .162 .154 .148	cm²/g 0, 520 258 190 168 158 151 146 138	

Table 5. Carbon

	Scatt	Scattering		dectric	Total experiment		
Photon energy	With coherent	With- out coherent	Calcu- lation K & L shells	Experi- ment	With coherent	With- out coherent	
Mer 0.01 .015 .02 .03 .04 .05 .06 .08	Barns/ atom 6, 88 5, 30 4, 64 4, 04 3, 71 3, 50 3, 37 3, 18 3, 02	Barns/ atom 3, 84 3, 77 3, 71 3, 58 3, 47 3, 37 3, 28 3, 10 2, 96	Barns/ atom 34, 2 9, 20 3, 60 0, 95 36 17 , 093 , 036	Barns/ atom 37, 3 9, 75 3, 82 1, 00 0, 37 , 17 , 083 , 036 , 017	cm ³ /g 2. 22 0. 755 . 424 . 253 . 205 . 184 . 174 . 161 . 152	cm²/g 2.06 0.678 .378 .230 .193 .178 .169 .157	

^{*} Kedge for A.

b Kedge for Cr.
c Kedge for Co.
d Kedge for Ni.
c Kedge for Cu.
f Kedge for Zn.

Table 6. Nitrogen

	Scat	Scattering		electric	Total experiment		
Photon energy	With coherent	With- out coherent	Calculation K & L shells	Experí- ment	With coherent	With- on coherent	
Mer		Barns/ atem	Parus) atom	Rarns/	cm²/g	cm²/g	
0.01	8, 96	4.48	70. 6	77.0	3, 70	3, 50	
.015	6, 72	4,40	19. 2	20, 4	1.17	1.07	
.02	5.73	1.33	7, 57	8, 02	0.591	0.531	
, 03	4. 84	4.18	2, 02	2.12	. 299	. 271	
, 04	4, 45	4.05	0, 77	0.79	. 225	. 208	
, 05	4, 14	3, 93	. 37	. 37	, 191	. 185	
iH),	3.98	3,82	. 20	. 20	, 180	. 173	
.08	3.73	3, 62	.08	.08	. 164	. 159	
. 10	3, 54	3, 45	04	. 04	. 151	. 170	

Table 7. Oxygen

Photon energy	Scattering		Photoc	electric	: Total experiment		
	With coher- ent	With- out coher- ent	Calculation K & L shells	Experi- ment	With coherent	Withou coheren	
	Parus/	Parus/	Parns!	Barns/			
Mer	atom	atom	atom	atom	CH12/41	* cm2/g	
0. 01	11.5	5, 12	130	142	5.78	5, 54	
. 015	8, 28	5, 03	36, 0	38. 2	1,75	1, 63	
. 02	6, 95	4, 94	14. 2	14.9	0, 823	0.747	
. 03	5, 77	4,78	3, 85	4, 04		32	
.04	5, 18	4, 62	1, 48	1.48	. 251	. 230	
. 05	4, 80	4, 49	0, 70	0.70	. 207	, 195	
, (H)	4.61	4, 37	. 39	. 39	188	. 179	
, 08	4, 30	4, 14	. 15	. 15	168	. 162	
. 10	4, 06	3, 94	.07	.07	. 156	151	

Table 8. Sodium

Photon energy	Scattering		Photoe	lectric	Total experiment		
	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment		Without coherent	
	Parus!	Barns/	Parus!	Parns!			
Mer	atom	atom	atom	atom	: cm2/g	cm ² /g	
0, 01	20	7, 04	530	572	15, 5	15.2	
. 015	14	6, 92	155	164	4, 66	4, 48	
. 02	11.2	6,80	62, 6	65, 7	2, 01	1,90	
. 03	8, 8	6, 57	17. 1	17.8	0.697	0,638	
. 04	7.8	6, 36	6, 7	6, 7	380	. 342	
. 05	7. 1	6, 17	3, 2	3, 2	270	. 245	
, 06	6, 67	6, 01	1,81	1, 81	. 222	. 205	
.08	6, 08	5, 69	0.72	0.72	. 178	. 168	
. 10	5, 70	5, 42	34	. 34	. 158	1. 151	

Table 9. Magnesium

Photon energy	Scattering		Photoc	dectric	Total experiment		
	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent	
Mer 0, 01 , 015 , 02 , 03 , 04 , 05 , 06 , 40	Rarns/ atom 25 17 13 10, 2 8, 7 7, 9 7, 4 6, 96 6, 24	Rarns/ atom 7, 68 7, 55 7, 42 7, 16 6, 94 6, 73 6, 55 6, 55 5, 91	Parns/ atom 763 226 92, 3 25, 7 10, 1 4, 9 2, 7 1, 08 0, 52	Barns/ atom 826 239 96, 3 27, 8 10, 1 4, 9 2, 7 1, 08 0, 52	cm ² /g 21.1 6.34 2.71 0.941 .466 .317 .250 .192 .167	cm ² /g 20, 6 6, 11 2, 57 0, 866 422 , 288 , 229 , 180 , 150	

TABLE 10. Aluminum

	Scatt	Scattering		electric	Total experiment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent	
Mer 0, 01 , 015 , 02 , 03	Barns/ atom 29 19 15	Barns/ atom 8, 32 8, 18 8, 03 7, 76	Barns/ atom 1, 060 316 131 36, 8	######################################	cm ² /g 26, 2 7, 91 3, 37 1, 11	cm²/g 25, 8 7, 67 3, 22 1, 03	
. 04 . 05 . 06 . 08 . 10	9. 8 8. 8 8. 1 7. 26 6. 79	7. 51 7. 29 7. 10 6. 72 6. 41	14. 5 7. 0 3. 9 1. 57 0. 77	14. 5 7. 0 3. 9 1. 57 0. 77	0. 543 . 353 . 268 . 197	0. 492 .319 .246 .185	

Table 11. Silicon

	Scattering		Photoel	ectric	Total experiment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent	
Mer	Barns/		Barns/	Barns/ alom	cm²/q	cm²/g	
0, 01	33	8, 96	1,440	1, 555	34.1	33.6	
. 015	22	8, 81	134	460	10.3	10, 1	
. 02	17	8, 65	181	188	4.40	4, 22	
. 03	12.8	8, 36	51.4	53. 5	1.42	1.33	
. 04	10.8	8, 09	20.4	20.4	0, 670	0.610	
. 05	9. 7	7.85	9, 9	9. 9	. 421	. 381	
. Oti	8.9	7.64	5, 6	5, 6	.311	. 284	
.08	8.0	7, 24	2, 2	2. 2	. 219	. 203	
. IO	7.38	6, 90	1.09	1,09	. 182	. 172	

Table 12. Phosphorus

	Scatt	ering	Photoel	ectric	Total experiment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent	
	Barns/	Barns/	Barns/	Barns/			
Mer	. atom	atom	atom	atom	cm²/g	cm2/q	
0, 01	38	9, 60	1, 900	2,047	40, 6	40.0	
. 015	25	9, 44	573	605	12, 3	12, 0	
. 02	. 19	9, 27	242	251	5, 25	5, 06	
. 03	, 14.3	8, 96	70, 1	72.6	1, 69	1, 59	
. 04	12.0	8,67	27.4	27.4	0, 766	0.702	
. 05	10, 6	8, 42	13.3	13.3	. 465	. 422	
. 06	9. 7	. 8, 19	7.6	7.6	. 336	. 307	
. 08	8, 6	7, 76	3, 0	3, 0	. 226	. 209	
. 10	7, 98	7, 39	1, 52	1, 52	. 185	. 173	

Table 13. Sulfur

	Scattering		Photoc	lectric	Total experiment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shelis	Experi- ment	With coherent	Without coherent	
	Barns/	Barns/	Barns/	Barns/			
Mer 0, 01	atom 44	atom	alom	atom 2,670	cm²/g 51, 0	Cm2/g	
0.015	20	10, 24 10, 06	2, 480 750	800	15.6	1 50, 4 1 15, 2	
. 02	22	9, 89	321	332	6, 65	6, 42	
. 03	15.9	9. 55	93. 1	96.2	2.11	1, 99	
. 04	13. 2	9, 25	36, 7	36, 7	0.938	0.863	
. 05	11.6	8, 98	17. 9	17, 9	. 554	, 505	
, OG	10, 7	8, 74	10, 3	10, 3	. 395	358	
. 08	9, 3	8, 27	4.1	4.1	. 252	, 232	
. 10	8, 6	7.89	2.1	2.1	. 201	. 188	

Photon energy	Scattering		Photoc	electric	Total ex	otal experiment		
	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With	Without coherent		
Mer 0,01 - 015 - 02 - 03 - 04 - 05 - 06 - 08 - 10	Barns/ atom 56 36 28 19 15, 8 13, 6 12, 4 10, 8 9, 85	Barns/ atom 11, 52 11, 32 11, 12 10, 75 10, 40 10, 10 9, 83 9, 31 8, 87	Rarus/ atom 3, 950 1, 230 527 156 62, 0 30, 5 17, 5 7, 1 3, 59	Barns/ atom 4, 230 1, 230 543 161 62, 0 30, 5 17, 5 7, 1 3, 59	cm ² /g 64, 6 20, 0 8, 61 2, 71 1, 17 0, 665 451 270 203	cm²/g 64. 0 19. 6 8. 36 2, 59 1, 09 0, 612 412 247 188		

TABLE	15.	Potassium .

	Scatt	ering	Photoel	eetri e	Total experiment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Withou coheren	
	Barns/	Baras	Barns!	Parns/		}	
Mer	atom	atom	atom	atom	cm^2/q	Cm2/g	
0, 01	63	12, 16	1,870	5, 208	81. 2	80. 1	
. 015	40	11, 95	1,542	1,620	25. 6	25. 1	
. 02	31	11.71	658	674	10.9	10.6	
. 03	21	11, 34	196	241-2	3, 44	3, 29	
.01	17. 1	10, 98	78, 4 :	80, 0	1, 50	1.40	
. 05	14.7	10, 66	38, 9	38. 9	0,826	0, 764	
, 06	13. 3	10, 37	22.3	22. 3	. 549	503	
. 08	11,6	9,82	9,0	9. 0	. 317	, 290	
. 10	10.5	9, 37	4.5	4. 5	. 231	. 214	

Table 16. Calcium

	Scatt	Scattering		ectric	Total experiment		
Photon energy	With coher- nt	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent	
Mer	Barns/	Barns:	Barns' alom	Barns/ atom	cm ² /q	n*/	
0.01	69	12.80	5,920	6, 330	96.2	cm [*] /q 95.3	
.015	41	12,58	1.890	1, 980	30.5	29, 9	
. 02	33	12, 36	Si.	827	12.9	12.6	
. 03	23	11, 94	243	250	4, 10	3, 94	
.01	18.5	11, 56	98. 4	101	1, 79	1, 69	
, 05	15.8	11, 22	49.0	49, 4	0.980	0, 911	
. (96	14.3	10, 92	28.1	28, 1	. 637	. 586	
. 08	12.3	10.34	11.4	11.4	356	. 327	
. 10	, 11.1	9, 80	5.9	5.9	256	. 237	

Table 17. Iron

	Scatt	Scattering		etrie	Total experiment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent	
Mer 0, 01 . 015	Barns/ atom 120 75	Rarns/ atom 16, 61 16, 35	Barns/ atom 15, 600 5, 130	Barns/ atom 16, 600 5, 335	cm²/g 180 58, 4	cm ² /g 179 57, 7	
. 02 . 03 . 04 . 05	37 20 24	16, 07 15, 52 15, 03 11, 59	2, 280 706 300 152	2,300 718 306 152	25. 4 8. 15 3. 61 1. 90	25. 0 7. 91 3. 46 1. 80	
, 06 , 08 , 10	20.7 17.2 15.4	14, 20 13, 44 12, 82	89, 1 37, 5 18, 9	89. 1 37. 5 18. 9	1.18 0.590 .370	1.11 0.550 .342	

	Scattering		Photoe	lectrie	Total experiment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent	
	Rarns/	Barns/	Barns/	Barns!			
Mev	atom	atom	atom	atom	Cm2/g	Cm2/g	
0.01	150	18, 56	22, 500	23, 600	225	224	
. 015	96	18.24	7, 690	8, 000	76. 7	76.0	
. 02	70	17, 92	3, 460	3, 460	33, 5	33.0	
. 03	46	17.31	1,090	1, 100	10, 9	10.6	
, 04	35	16, 76	465	465	4,74	4, 57	
. 05	28	16, 27	238	238	2,52	2.41	
. 06	24	15, 83	141	141	1.56	1.49	
. 08	20, 2	14, 99	59, 7	59, 7	0, 757	9,708	
. 10	17. 9	14, 29	30, 5		459	, 425	

Table 19. Molybdenum

	Scattering		Photoc	lectric	Total experiment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L. & M shells	Experi- ment	With	Without coherent	
	Barns/	Rarns/	Parus	Barns)			
Mer	atom	atom	atom	atom	cm2/g	cm^2/g	
0, 01	340	26, 9	11,400	13, 400	86.3	84.3	
. 015	220	26, 4	3, 480	4, 370	28, 8	27. 6	
*, 0200	160	26, 0	1,510	1, 871	12.8	11.9	
.0200	160	26, 0	13, 000	13, 200	83, 9	83, 0	
. 03	98	25.1	4, 260	1, 360	28, 0	27. 5	

* K edge; at this and lower energies data for the L and M shells are given while at this and higher energies data for the L, M, and K shells are given,

Table 20. Tin

	Scattering		Photoc	dectric	Total experiment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells		With	Without coherent	
	Barns/	Barns/	Rarns/	Barns/	:		
Mer	atom	afom	atom	atom	cm² g	cm2/g	
0.01	510	32, 0	24,000	28, 500	147	145	
. 015	340	31.4	7,410	9,560	47. 7	46, 1	
. 02	240	30, 9	3, 220	4, 040	21.7	20, 7	
n, 02919	150	30, 0	1,050	1,4(8)	7,87	7, 26	
.02919	150	30, 0	8, 580	9, 000	46, 4	45.8	
. 03	140	29, 8	8, 150	8, 220	42.4	41.9	

* K edge; at this and lower energies data for the L and M shells are given while at this and higher energies data for the L, M, and K shells are given.

Table 21. Indine

	Scattering		Photoe	dectrie	Total ex	periment		
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent		
Mer	Rarns/	Rarns/	Rarus) atom	Rarnsi atom	cm250			
0, 01	500	33, 9	29,800	37, 700	182	Cm²/q 179		
.015	380	33, 3	9, 360	11,900	58.3	56.6		
. 02	270	32.8	4, 130	5, 300	26, 9	25.7		
. 03	160	31.6	1,260	1, 760	9.11	8, 50		
*, 03316	150	31.3	933	1, 340	1, 07	6, 51		
.03316	150	31.3	7, 510	7, 510	36, 4	35, 8		

* K edge; at this and lower energies data for the L and M shells are given while at this and higher energies data for the L, M, and K shells are given.

Table 22, Tungsten

	Scattering		Photoc	lectric	Total experiment		
Photon energy	With coher ent	With- out coher- ent	Calculation K, L, & M shelis	Experi- ment	With coherent	Without	
•	Parns	 Parusi	· Barus'	Barns/		! ·• · · ·	
Mer	atom	atom	atom	atom	: cm²/g	cm²/a	
0, 91	1, 300	47, 4	17, 7(K)	27, 200	93, 4	89.3	
01020	1, 200	47. 3	16,800	26, 000	89, 0	85, 0	
6, 01200), (RN)	19, 9	61, 700	74, 700	248	245	
.015	840	46, 5	36, 000	42, 200	: 141	138	
62	590	45, 7	16,000	19, 900	67.0	65.2	
, 03	350	44, 2	5,040	6, 670	23.1	22.0	
.01	240	42.8	2, 220	3, 020	10.7	10,0	
. 05	180	11.5	1, 160	1,660	6, 03	5, 57	
.03	145	40, 4	674	984	3, 70	3, 36	
 0.7951 	122	39, 4	437	656	2, 55	2, 25	

** Lyedge; at this and lower energies data for the M shell are given, ** L1 edge; from this energy to the K-edge energy data for the L and M shells are given, ** K edge; at this and higher energies data for the L, M, and K shells are

Table 23. Platinum

Photon : energy	Scattering		Photoel	ectric	Total experiment	
	With coher- ent	With- out coher- ent	Calculation K. L, & M shells	Experi- ment	With coherent	Without coherent
	Barns!	Barnsi	Parns/	Parns!		
Mer	atom	atom	atom' '	atom	: cm2/g	cm2/g
0.01	. 1,400	19, 9	22,000	36, 000	115	111
4, 01156	1,200	49, 6	11,800	23, 400	76, 0	72.0
1.01387	1, (XX)	49.2	53, 900	61, 900	194	191
. 015	940	19, 1	43,800	50, 300	158	155
. 02	670	48, 2	19,700	23, 800	75.5	73. 6
. 03	4(10)	46.6	6, 240	8, 100	26, 2	25.1
, 04	240	45. 1	2,720	3, 770	12.5	11.8
0.5	210	: 43. S	1, 110	2,040	6,91	5, 43
.03	163	42.6	836	1, 2!7	4, 26	3, 89
5, 07838	117	40, 6	380	560	2, 09	1.85

* L_{Y} edge; at this and lower energies data for the M shell are given. * L_{Y} edge; from this energy to the K edge energy data for the L_{Y} and M shells are given. * K edge; at this and higher energies data for the L_{Y} M_{Y} and K shells are given.

Table 24. Thallium

Photon energy	Scattering		Photoe	lectric	Total experiment	
	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent
Mer 0, 01 *, 01266 5, 01535 , 02 , 03 , 03 , 05 , 05 , 08 *, 08552 , 08552	Rarns/ atom 1,500 1,200 990 730 430 300 220 180 124 114	Barns/ atom 51. 8 51. 3 50. 7 50. 1 48. 4 45. 8 45. 4 44. 2 41. 9 11. 3	Parns/ atom 23,000 13,400 47,200 22,700 7,220 3,200 1,660 976 420 341 2,577	Parns/ atom 44, 200 24, 600 54, 000 57, 100 9, 840 4, 580 2, 460 1, 480 622 530	cm ² /q 135 76 162 82, 0 30, 3 14, 4 7, 9 4, 9 2, 2 1, 9 7, 93	cm ² /g 130 73 159 80, 0 29, 2 13, 6 7, 4 4, 5 2, 0 1, 7 7, 72

• L_1 edge; at this and lower energies data for the M shell are given, b. L_2 edge; from this energy to the K-edge energy data for the L and M shells are given. • K edge; at this and higher energies data for the L, M, and K shells are given.

Table 25. Lead

Photon energy	Scattering ,		Photoel	ectric	Total experiment	
	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent
	Barns/	Barns/	Barns/	Barns/		
Mer	atom	atom	alom	utom	cm²/g	cm²/g
0.01	1,600	52.5	27, 500	47, 200	142	137
*. 01304 b. 01587	1,200	51.8 51.3	13, 200	23, 200	71.0	67.6 168
.02	750	50.7	45, 400 24, 000	57, 800 30, 900	92	90
. 03	450	49.0	7, 620	10, 500	31.8	30.6
.04	310	47.4	3, 310	4, 880	15.1	14.3
.05	230	46.0	1,740	2,690	8, 49	7, 96
. 06	180	44.8	1,040	1,580	5.12	4.72
. 08	127	42.4	444	688	2.37	2.12
c, 08800	113	11.6	331	523	1.85	1.64
. 08800	113	41.6	2,510		7, 63	7.42

* L_2 edge; at this and lower energies data for the M shell are given, b L_1 edge; from this energy to the K-edge energy data for the L and M shells are given, c K edge; at this and higher energies data for the L, M, and K shells are given.

given.

Table 26, Uranium

	Scattering		Photoc	electric	Total experiment	
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent
	Parus/	Barns/	Parns/	Parns/		
Mer	atom	atom	atom	atom	cm²/g	cm²/g
0. 01	2, 100	58, 9	44,600	94, 400	244	239
.015	1,400	57. 9	14,500	33, 800	89, 1	85.7
3,01716	1, 200	57, 4	10,000	23, 800	63.4	60, 4
b, 02175	880	56, 5	29, 400	46, 100	119	117
.03	5(0)	54.9	12,000	20, 300	52.9	51.5
. 04	400	53. 2	5, 250	9, 910	26. 1	25, 2
. 05	300	51.6	2, 780	5, 666	15.1	14.5
. 06	230	50, 2	1,640	3, 550	9, 57	9.11
. 08	163	47.6	716	1,700	4.71	4, 42
. 10	123	45.3	374	948	2.71	2, 51
c, 1156	103	43, 8	239	640	1.88	1.73
. 1156	103	43, 8	3, 790		4, 79	4, 64

 $^{\circ}$ L_{1} edge; at this and lower energies only the M shell data are given, $^{\circ}$ L_{1} edge; from this to the K-edge energy data for the L and M shells are given, $^{\circ}$ K edge; at this and higher energies data for the L, M, and K shells are given.

given.

Table 27. Water

	Scattering		Photoc	lectric	Total experiment	
Photon energy	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent
Mer	Barns/	Barns/	Barns/ atom	Barns/ atom	cm²/a	cm²/g
0. 01	12.8	6, 40	130	142	5, 18	4, 96
. 015	9, 54	6, 29	36.0	38, 2	1, 60	1, 49
. 02	8, 19	6, 18	14. 2	14, 9	0.772	0, 705
. 03	6, 96	5, 97	3, 85	4, 04	. 368	, 335
. 04	6.34	5.78	1,48	1, 48	. 262	, 243
, 05	5, 92	5, 61	0.70	0, 70	. 221	. 211
. (%)	5, 70	5.46	.39	. 39	. 204	. 196
. 08 . 10	5, 33 5, 05	5, 17 4, 93	. 15	. 15	. 183	. 178 . 167

Table 28. Sodium Iodide

Photon energy	Scattering		Photoc	dectric	Total experiment	
	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With	Without coherent
Mer 0.01 .015 .02 .03 *.03316 .03316 .04 .05 .06 .08	Barns! atom 610 390 280 170 160 160 130 96 79 60	Barns/ atom 41, 0 40, 3 39, 6 38, 2 37, 8 37, 0 35, 9 34, 9 34, 9 31, 5	Parns! atom 30, 300 9, 520 4, 190 1, 280 945 7, 520 4, 500 2, 470 1, 500 678 360	Rarns/ atom 38,300 12,100 5,460 1,780 1,350 7,520 4,500 2,470 1,500 678 360	cm ² /g 156 50, 2 2, 1 7, 84 6, 07 30, 9 18, 6 10, 3 6, 35 2, 97 1, 65	cm²/g 154 48, 8 22, 1 7, 31 5, 58 30, 4 18, 2 10, 1 6, 17 2, 86 1, 57

^{*} K edge of Iodine; at this and lower energies data for the L and M shells are given while at this and higher energies data for the L, M, and K shells are given.

Table 29. Calcium Phosphate

Photon energy	Scattering		Photoc	electric	Total experiment	
	With coher- ent	With- out coher- ent	Calculation K, L, & M shells	Experi- ment	With coherent	Without coherent
Mev 0. 01 . 015 . 02 . 03 . 04 . 05	Barns/ atom 375 248 193 144 121 107 99, 2	Barns/ atom 98, 6 96, 9 95, 2 91, 9 89, 0 86, 4 84, 1	Rarns/ atom 22, 600 7, 100 3, 030 900 362 179 103	Parns/ atom 24, 200 7, 460 3, 100 928 370 180 103	cm ² /g 47. 7 15. 0 6. 40 2. 08 0. 954 . 557 . 393	cm ² /g 47, 2 14, 7 6, 21 1, 98 0, 891 517

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Table 30. Air 0.755 N, 0.232 O, 0.013 A by weight

Photon	To	in1	Photon	Total		
energy	With coherent	Without coherent	energy	With coherent	Withou coheren	
Mev	c m²/g	cm2/g	Mer	cm²/g	cm²/q	
0.003	165		0.015	1.55	1.44	
*, (¥)32	135	1	. 020	0.749	0, 683	
.0032	157	1	. 030	. 347	315	
. 094	83.4	1	. 040	. 243	. 225	
. 005	42.1	1	. 050	. 203	. 193	
, (X)G	23. 5	1	. 060	185	. 177	
.008	9, 96	1	, 080	. 166	. 161	
.010	4, 97	4, 76	. 10	. 155	. 151	

[&]quot; K edge for Argon.

Table 31. Concrete

0.59% H, 49.83% O, 31.58% Si, 4.56% Al, 8.26% Ca, 1.22% Fe, 0.24% Mg, 1.71% Nu, 1.92% K, 0.12% S by Weight, Density=2.35 g/cm³.

Photon cnergy Inco- herent scatter- ing a	Inco-	Total b		Pho- ton ener- gy	Inco- herent scatter- ing *	Total b		
	With coher- ent	Without coher- ent	With coher- ent			Withou coher- ent		
Mev	cm²/g	cm²/g	cm²,g	Men	cm²/g	cm²/g	cm²/g	
0.01	0.193	26.9	26. 5	1.0	0.0637	0.0637	0.0637	
.015	.190	8.32	8, 12	1.5	. 0517		.0519	
. 02	. 186	3, 59	3.45	2.0	.0441		.0447	
. 03	. 180	1.21	1.13	3.0	. 0347		.0364	
. 04	. 174	0. 593	0, 548	4.0	. 0289		.0318	
. 05	. 169	. 389	. 351	5, 0	. 0250		. 0288	
. 06	. 165	. 288	. 268	6,0	. 0221		, 0268	
. 08	. 156	. 210	. 198	8, 0	.0181		.0242	
. 10	. 149	.178	. 170	10	.0154		.0228	
. 15	. 134	.144	. 139	15	.0114		.0211	
. 20	. 123	. 127	. 125	20	.00912		.0206	
. 30	.107	. 108	. 107	30	.00663		. 0205	
. 40	0955	. 0962	.0957	40	. 00526		. 0209	
, 50	. 0872	.0875	.0872	50	.00439		.0213	
. 60	.0807	. 0808	. 0807	60	.00378		.0217	
. 80	. 0709	.0710	. 0709	80 100	.00298		.0225	

^{*} Incoherent scattering is given by the Klein-Nishina formula for free

electrons.

b The total is compiled from the last two columns in the tables for the elements.

THE NATIONAL BUREAU OF STANDARDS

The scope of activities of the National Bureau of Standards at its major laboratories in Washington, D.C., and in Boulder, Colorado, is suggested in the following listing of the divisions and sections engaged in technical work. In general, each section carries out specialized research, development, and engineering in the field indicated by its title. A brief description of the activities, and of the resultant publications, appears on the inside of the front cover.

WASHINGTON, D.C.

Electricity and Electronics. Resistance and Reactance. Electron Devices. Electrical Instruments. Magnetic Measurements. Dielectrics. Engineering Electronics. Electronic Instrumentation. Electrochemistry.

Optics and Metrology. Photometry and Colorimetry. Optical Instruments. Photographic Technology. Length. Engineering Metrology.

Heat. Temperature Physics. Thermodynamics. Cryogenic Physics. Rheology. Molecular Kinetics. Free Radicals Research.

Atomic and Radiation Physics. Spectroscopy. Radiometry. Mass Spectrometry. Solid State Physics. Electron Physics. Atomic Physics. Neutron Physics. Radiation Theory. Radioactivity. X-ray. High Energy Radiation. Nucleonic Instrumentation. Radiological Equipment.

Chemistry. Organic Coatings. Surface Chemistry. Organic Chemistry. Analytical Chemistry. Inorganic Chemistry. Electrodeposition. Molecular Structure and Properties of Gases. Physical Chemistry. Thermochemistry. Spectrochemistry. Pure Substances.

Mechanics. Sound. Mechanical Instruments. Fluid Mechanics. Engineering Mechanics. Mass and Scale. Capacity, Density, and Fluid Meters. Combustion Controls.

Organic and Fibrous Materials. Rubber. Textiles. Paper. Leather. Testing and Specifications. Polymer Structure. Plastics. Dental Research.

Metallurgy. Thermal Metallurgy. Chemical Metallurgy. Mechanical Metallurgy. Corrosion. Metal Physics.

Mineral Products. Engineering Ceramics. Glass. Refractories. Enameled Metals. Constitution and Microstructure.

Building Technology. Structural Engineering. Fire Protection. Air Conditioning, Heating, and Refrigeration. Floor, Roof, and Wall Coverings. Codes and Safety Standards. Heat Transfer. Concreting Materials.

Applied Mathematics. Numerical Analysis. Computation. Statistical Engineering. Mathematical Physics.

Data Processing Systems. SEAC Engineering Group. Components and Techniques. Digital Circuitry. Digital Systems. Applications Engineering.

• Office of Basic Instrumentation.

Office of Weights and Measures.

BOULDER, COLORADO

Cryogenic Engineering. Cryogenic Equipment. Cryogenic Processes. Properties of Materials. Gas Lique-faction.

Radio Propagation Physics. Upper Atmosphere Research. Ionosphere Research. Regular Prediction Services. Sun-Earth Relationships. VHF Research. Radio Warning Services. Airglow and Aurora. Radio Astronomy and Arctic Propagation.

Radio Propagation Engineering. Data Reduction Instrumentation. Radio Noise. Tropospheric Measurements. Tropospheric Analysis. Propagation-Terrain Effects. Radio-Meteorology. Lower Atmosphere Physics.

Radio Standards. High-Frequency Electrical Standards. Radio Broadcast Service. Radio and Microwave Materials. Electronic Calibration Center. Microwave Circuit Standards.

Radio Communication and Systems. Low Frequency and Very Low Frequency Research. High Frequency and Very High Frequency Research. Modulation Systems. Antenna Research. Navigation Systems. Systems Analysis. Field Operations.